

# How Clouds Form: Water Vapor, Cooling, and Condensation



Roadmap: evaporation → water vapor → cooling → condensation onto nuclei



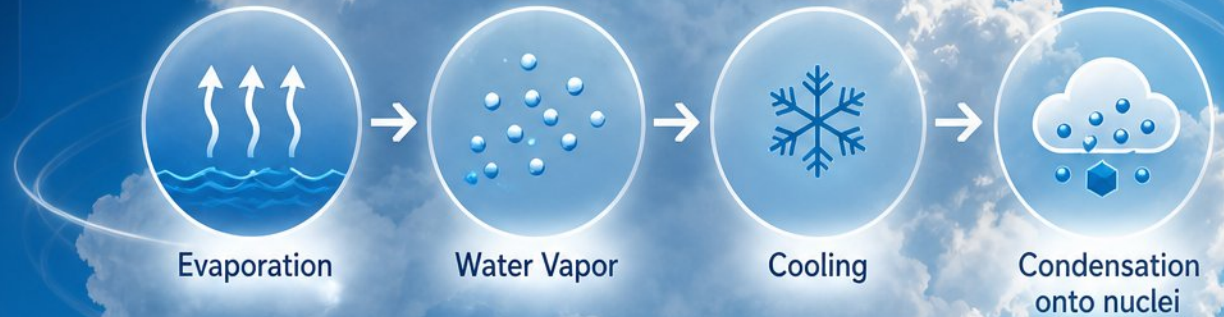
Explain how vapor cooling, CCN, and droplet/ice formation create clouds



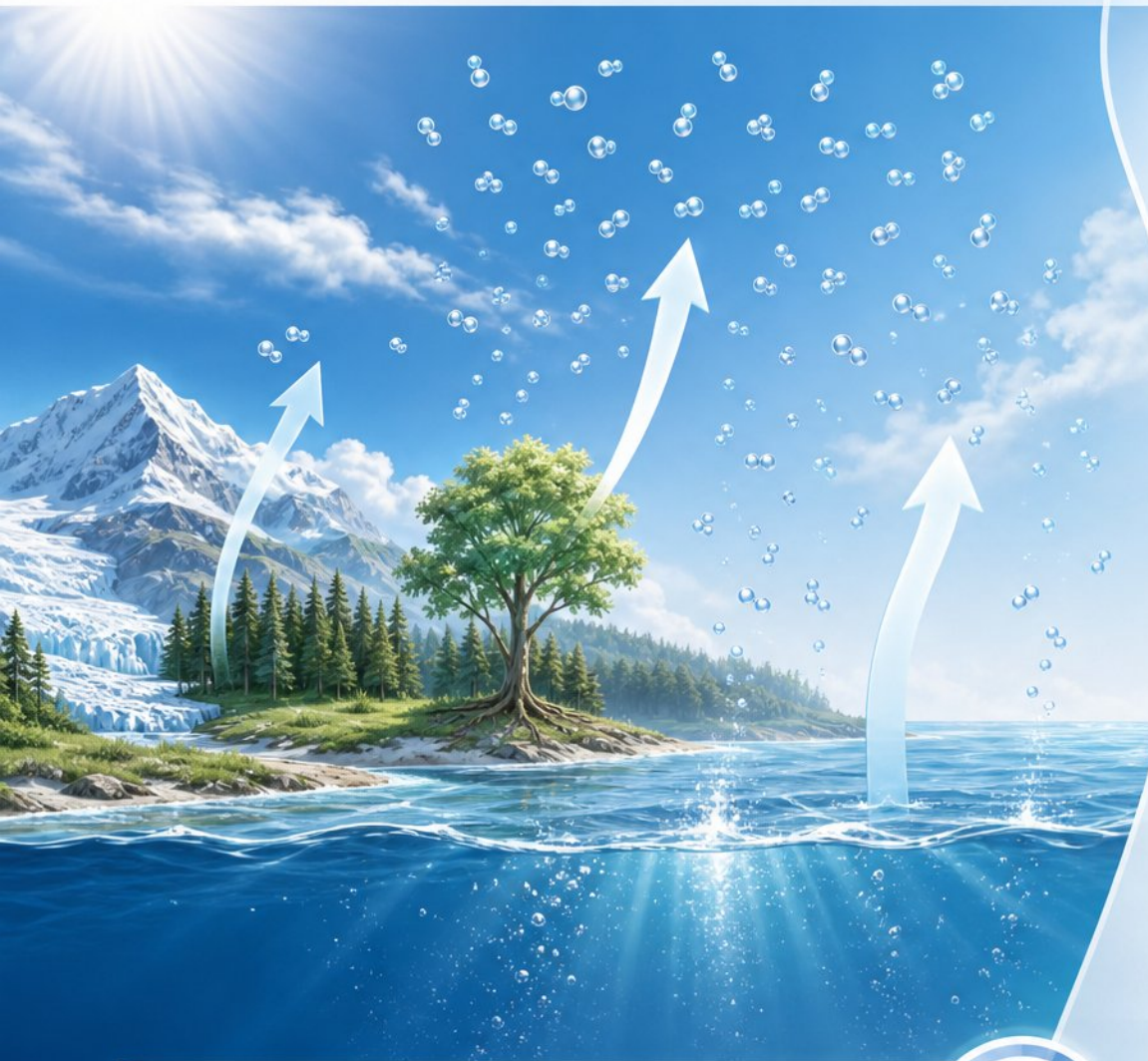
Applications: weather prediction, aviation, climate, sky observation



Misconceptions: clouds are not steam; condensation releases latent heat



# The Invisible Fuel: Water Vapor in the Atmosphere



- Water vapor enters the air via ocean/lake evaporation, plant transpiration, and ice/snow sublimation
- It is an invisible gas, distinct from visible liquid water droplets or ice crystals
- Vapor pressure ( $e$ ) measures water vapor's partial pressure; saturation vapor pressure ( $e_s$ ) is the max possible at a given temperature
- The Clausius-Clapeyron equation shows  $e_s$  increases rapidly with temperature—warmer air can hold much more moisture

**INVISIBLE**  
Water Vapor (Gas)

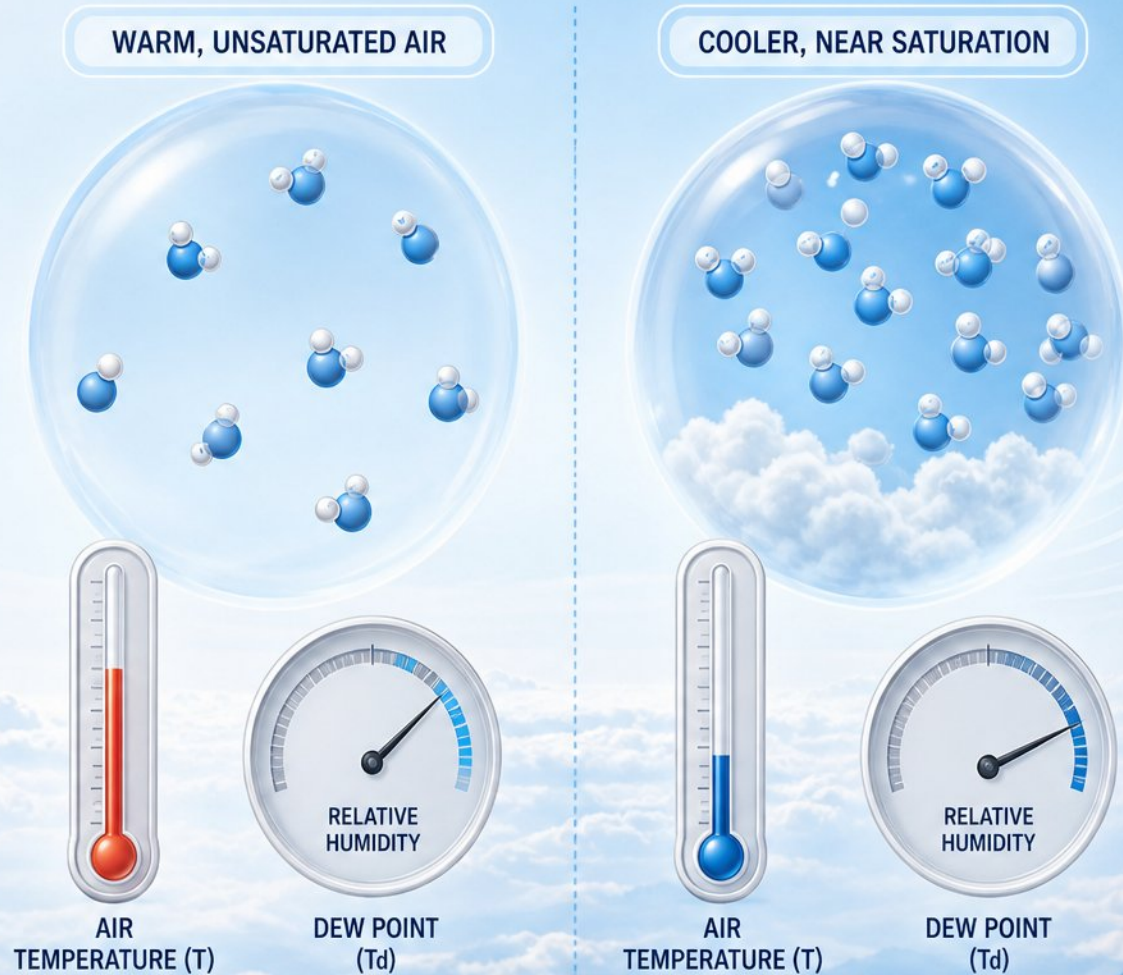
INVISIBLE  
TO  
VISIBLE



**VISIBLE**  
Liquid Droplets  
& Ice Crystals

# Measuring Moisture: Humidity and Dew Point

- **Absolute measures:** vapor pressure, mixing ratio, and specific humidity quantify actual water vapor
- **Relative Humidity** ( $RH = e/e_s \times 100\%$ ) indicates how close air is to saturation, not total water content
- **Dew point temperature ( $T_d$ ):** the temperature where saturation occurs—key to estimating cloud base
- **Dew point depression ( $T - T_d$ )** shows how much cooling is needed to reach saturation



# Cooling the Air: Pathways to Saturation

- - Rising air expands in lower pressure and cools adiabatically
- - Dry adiabatic lapse rate is  $\sim 9.8^{\circ}\text{C}$  per km for unsaturated air
- - Moist adiabatic lapse rate is  $\sim 5\text{--}6^{\circ}\text{C}$  per km after saturation
- - Lifting mechanisms: surface heating, orographic lift, frontal lifting, convergence

## Lifting Mechanisms



Surface heating



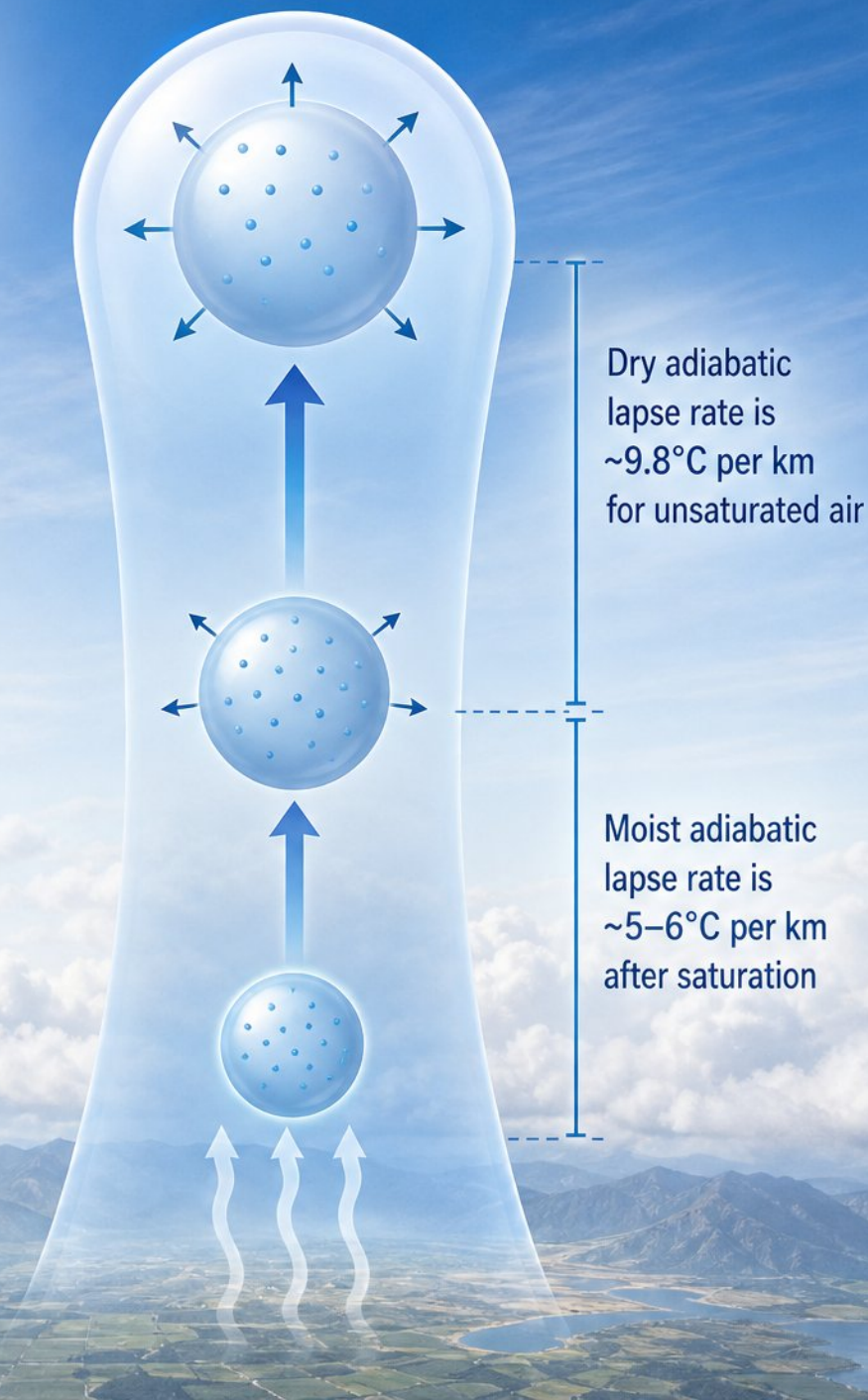
Orographic lift



Frontal lifting

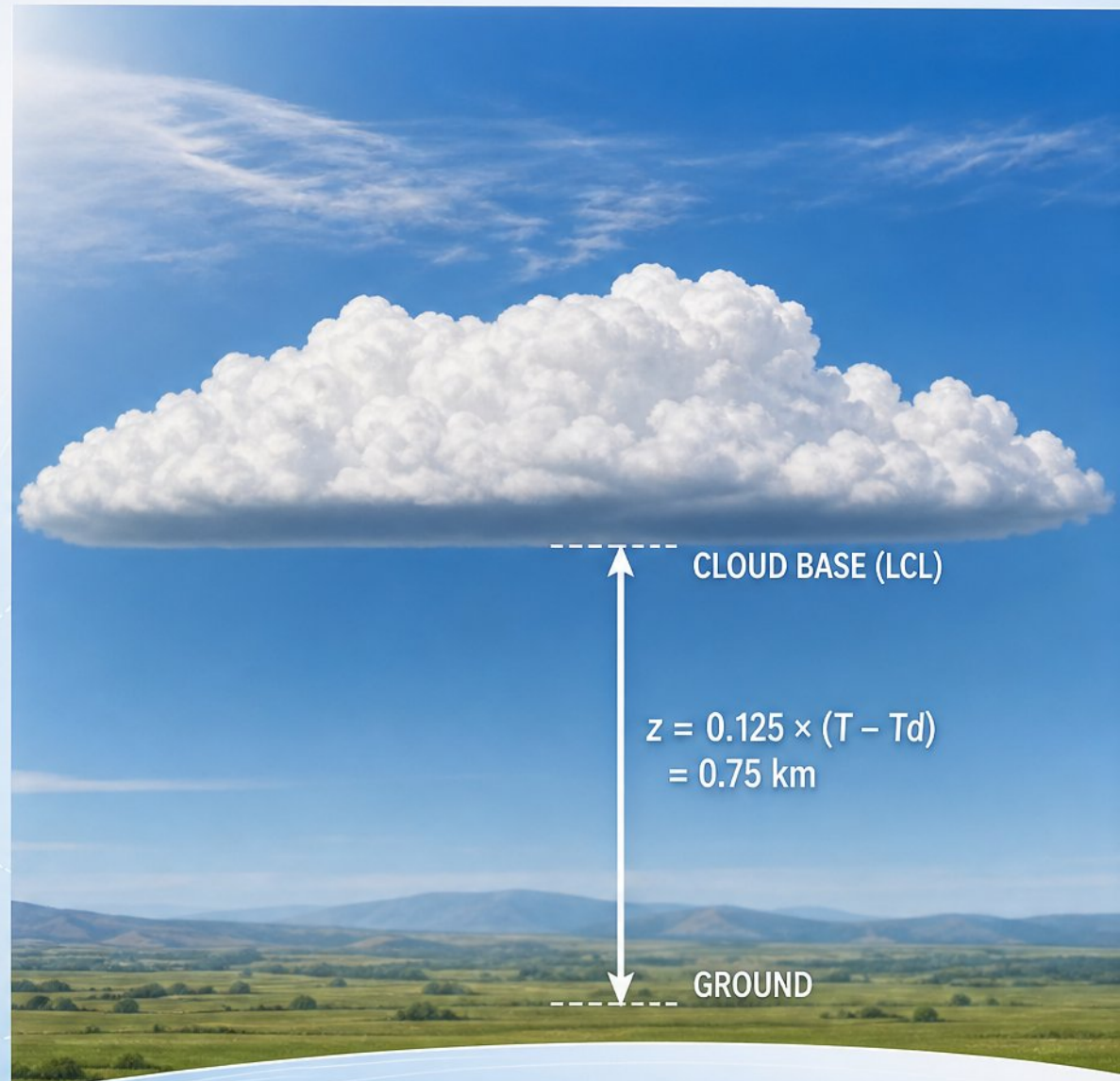
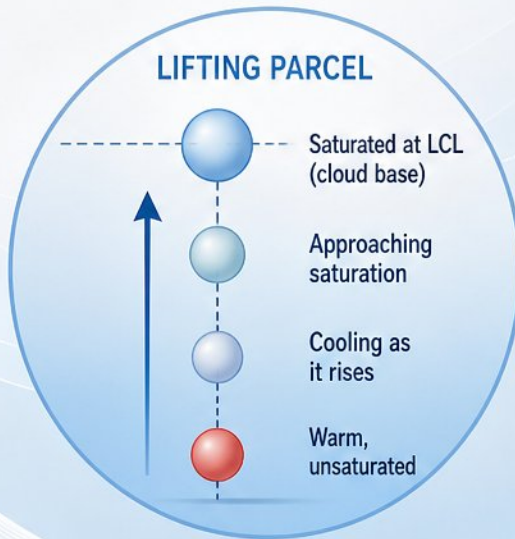


Convergence



# From Dew Point to Cloud Base: The Lifting Condensation Level

- LCL is the height where a rising air parcel cools to saturation
- Cloud-base formula:  $z = 0.125 \times (T - T_d) \text{ km}$
- Example:  $T = 20^\circ\text{C}$ ,  $T_d = 14^\circ\text{C}$  gives cloud base at 0.75 km
- Flat cumulus bases directly visualize the LCL altitude



# The Nucleation Problem: Why Water Vapor Needs a Surface

- **Homogeneous nucleation:** spontaneous droplet formation requires extreme supersaturation ( $RH \sim 300\%$ )—virtually impossible in real atmosphere
- **Heterogeneous nucleation:** water vapor condenses onto existing atmospheric particles called cloud condensation nuclei (CCN)
- **CCN** are a subset of aerosols, typically  $0.1\text{--}1\text{ }\mu\text{m}$  in diameter, that provide the necessary surface for condensation
- Every cloud droplet forms on a CCN; without nuclei, no visible clouds can exist



# Types and Sources of Cloud Condensation Nuclei



- Natural CCN: sea salt, dust, volcanic sulfates, biological particles, wildfire smoke



- Anthropogenic CCN: industrial sulfates, soot, secondary aerosols (gas-to-particle conversion)



- Hygroscopic (soluble) vs. hydrophobic particles; solubility enables droplet activation



- Continental CCN ( $\sim 1,000/\text{cm}^3$ ) yield more, smaller droplets than marine CCN ( $\sim 100/\text{cm}^3$ )

## CCN PARTICLES: TYPES AND SOURCES

### NATURAL ORIGIN

Sea Salt



Dust



Volcanic Sulfates



Biological Particles



Wildfire Smoke



### ANTHROPOGENIC ORIGIN

Industrial Sulfates



Soot



Secondary Aerosols  
(Gas-to-Particle Conversion)



Continental CCN  
( $\sim 1,000/\text{cm}^3$ )



Marine CCN  
( $\sim 100/\text{cm}^3$ )



# Köhler Theory: Curvature, Solutes, and Droplet Activation



- Kelvin (curvature) effect raises equilibrium vapor pressure over small droplets



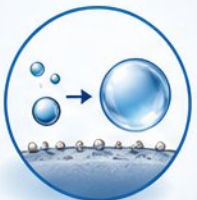
- Raoult (solute) effect lowers vapor pressure by dissolved CCN material



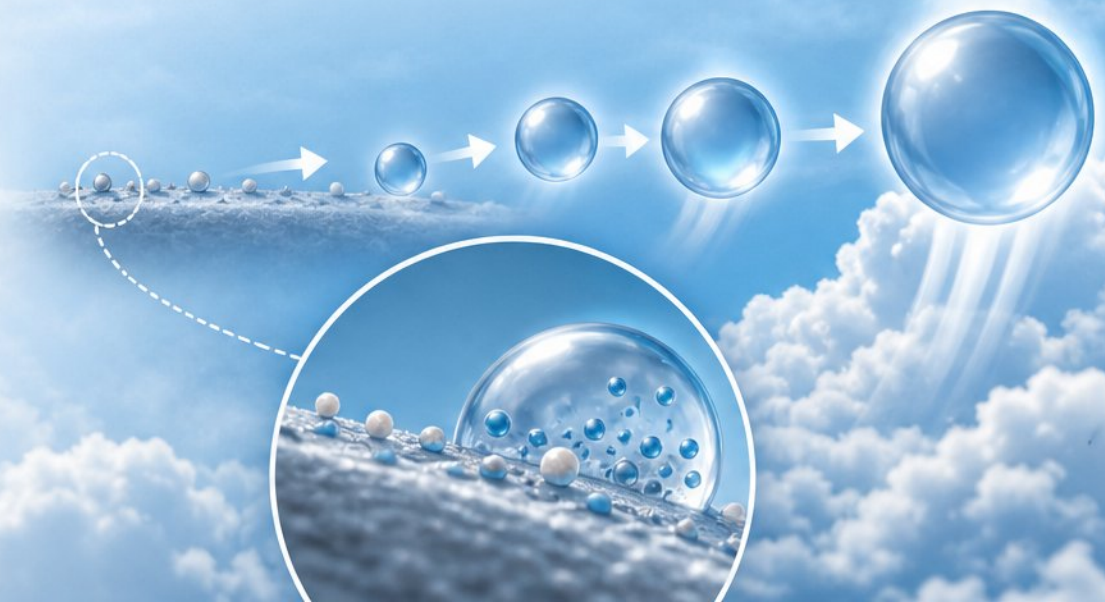
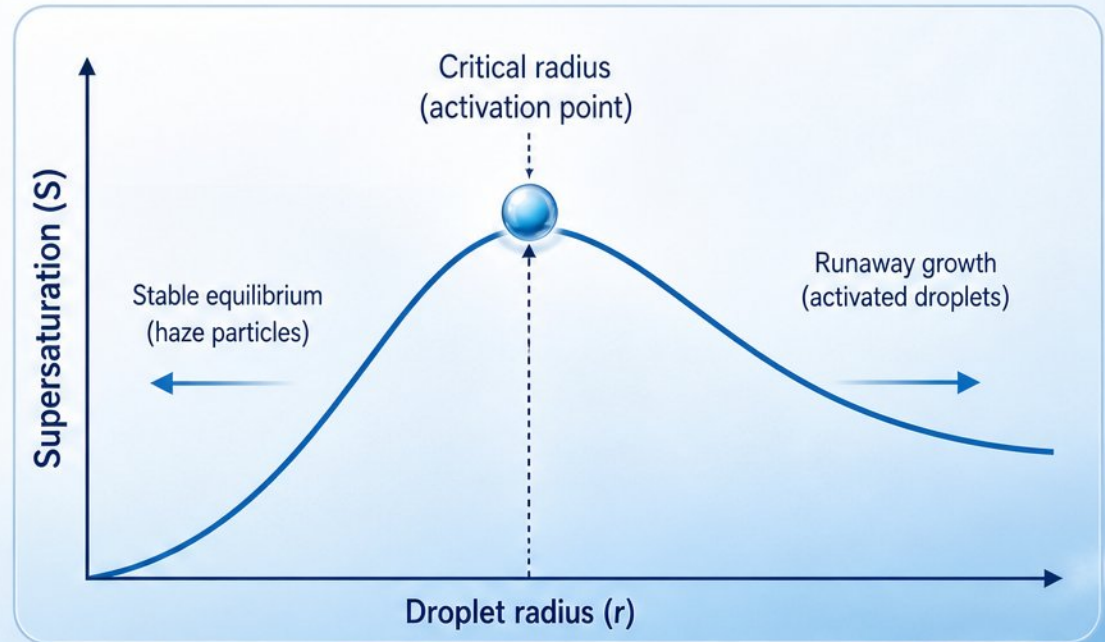
- The Köhler curve shows supersaturation vs. droplet radius; the peak defines the activation barrier



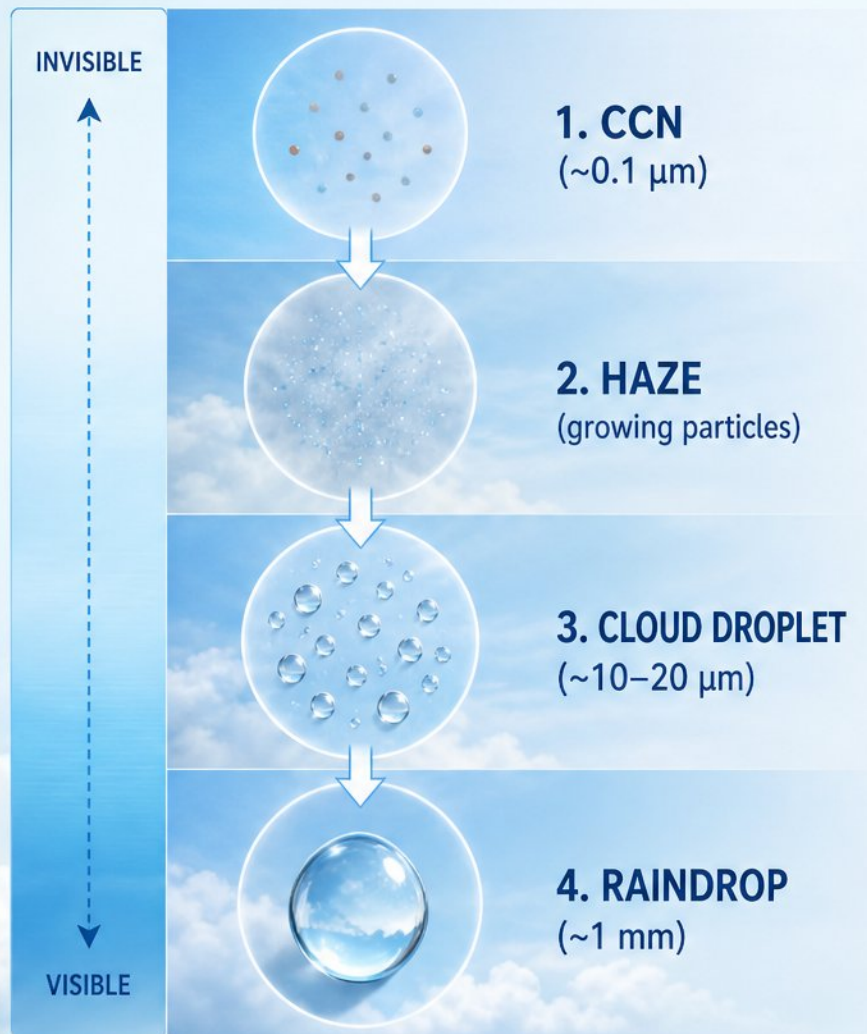
- Haze particles exist in stable equilibrium below 100% RH; activated droplets grow freely above the critical supersaturation



- Larger or more hygroscopic CCN activate first at lower supersaturations



# From Vapor to Droplet: The Condensation Process Step by Step



SIZE SCALE (INCREASING)

$\sim 0.1 \mu\text{m}$

$\sim 10\text{--}20 \mu\text{m}$

$\sim 1 \text{ mm}$



- Air rises  $\rightarrow$  expands  $\rightarrow$  cools  $\rightarrow$  RH reaches 100%  $\rightarrow$  water vapor condenses onto CCN



- Initial rapid growth by vapor diffusion; slows as droplets compete for water vapor



- CCN ( $\sim 0.1 \mu\text{m}$ )  $\rightarrow$  cloud droplet ( $\sim 10\text{--}20 \mu\text{m}$ )  $\rightarrow$  raindrop ( $\sim 1 \text{ mm}$ )



- Latent heat release warms the parcel, adding buoyancy and fueling continued lift



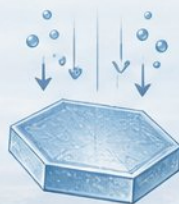
- Droplet fall speeds ( $\sim 1 \text{ cm/s}$ ) are far less than typical updrafts, keeping clouds aloft

# Below Freezing: Supercooled Water and Ice Nuclei

- Warm clouds (liquid only) vs. cold clouds (mixed-phase or fully glaciated)
- Supercooled water: liquid droplets persist down to  $\sim -40^{\circ}\text{C}$
- Ice nuclei (IN) are far scarcer than CCN, enabling supercooled clouds
- Modes of ice nucleation: deposition, condensation freezing, immersion, contact



Deposition



Condensation freezing



Immersion



Contact



# Ice Crystal Growth: The Wegener–Bergeron–Findeisen Process

- - Saturation vapor pressure over ice is lower than over liquid water at subfreezing temperatures.
- - In mixed-phase clouds, ice crystals grow at the expense of evaporating supercooled droplets.
- - The WBF process occurs when ambient vapor pressure falls between the ice and liquid saturation curves.
- - Results in rapid ice crystal growth to precipitation sizes, often leading to complete cloud glaciation.



# Ice Crystal Habits and Secondary Ice Production

## ICE CRYSTAL HABITS

Plates



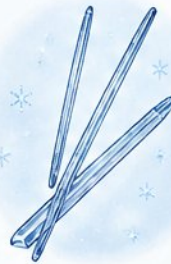
Columns



Dendrites



Needles



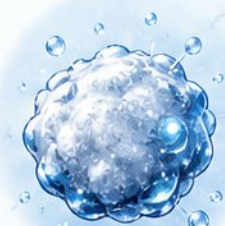
## AGGREGATION

Crystals Stick Together

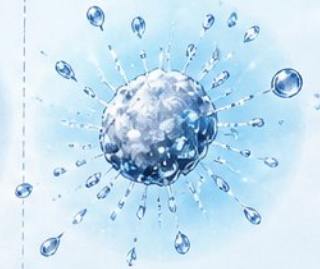


## RIMING & SECONDARY PROCESSES

Riming  
Graupel or Hail



Hallett-Mossop Process  
Rime Splintering



- Ice crystal shape depends on temperature and supersaturation: plates, columns, dendrites, needles



- Aggregation: crystals stick together near -5 °C, accelerating growth

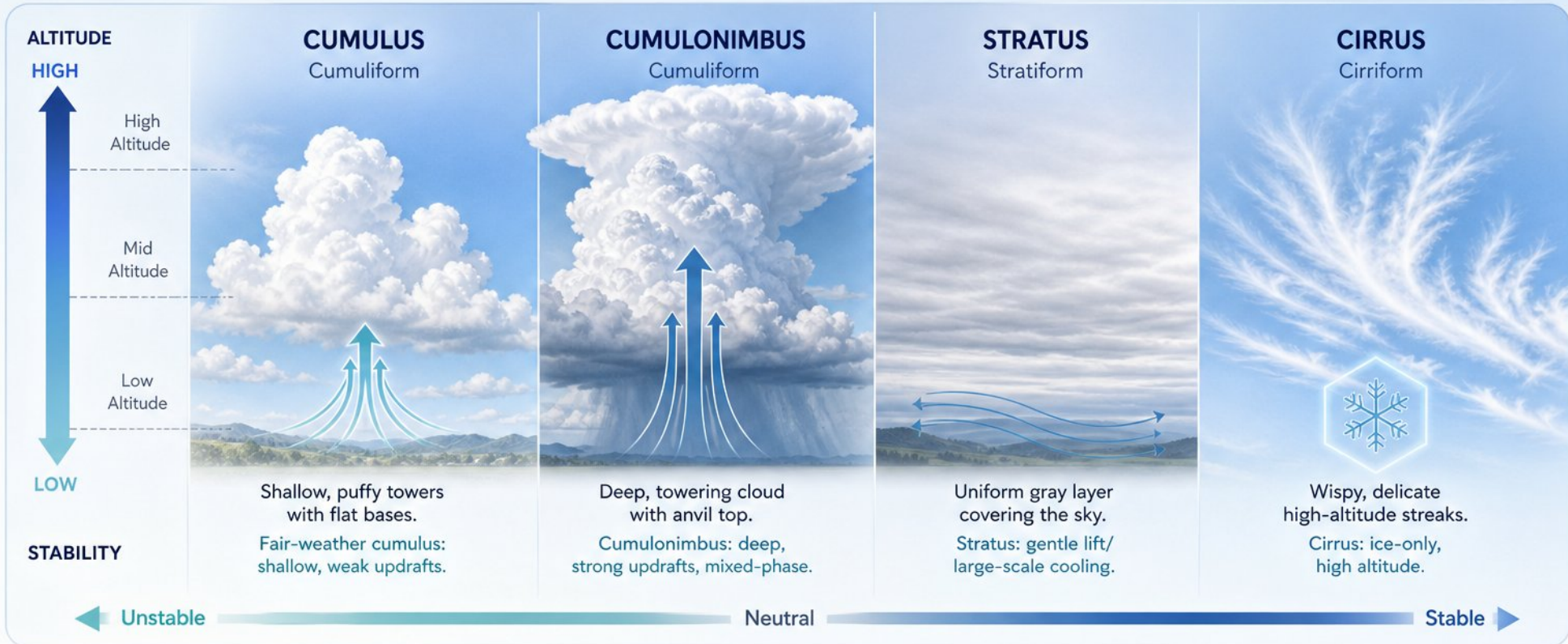


- Riming: supercooled droplets freeze on contact, forming graupel or hail



- Hallett-Mossop process: rime splintering at -3 to -8 °C multiplies ice crystal numbers

# Reading the Sky: What Clouds Tell Us About Atmospheric Processes



- Cumuliform = convection (vertical), Stratiform = layer (horizontal), Cirriiform = ice (wispy)
- Cloud type controlled by updraft strength, atmospheric stability, and moisture supply
- Fair-weather cumulus: shallow, weak updrafts. Cumulonimbus: deep, strong updrafts, mixed-phase
- Stratus: gentle lift/large-scale cooling. Cirrus: ice-only, high altitude
- Cumulus base = visual LCL; cloud top height reflects atmospheric stability



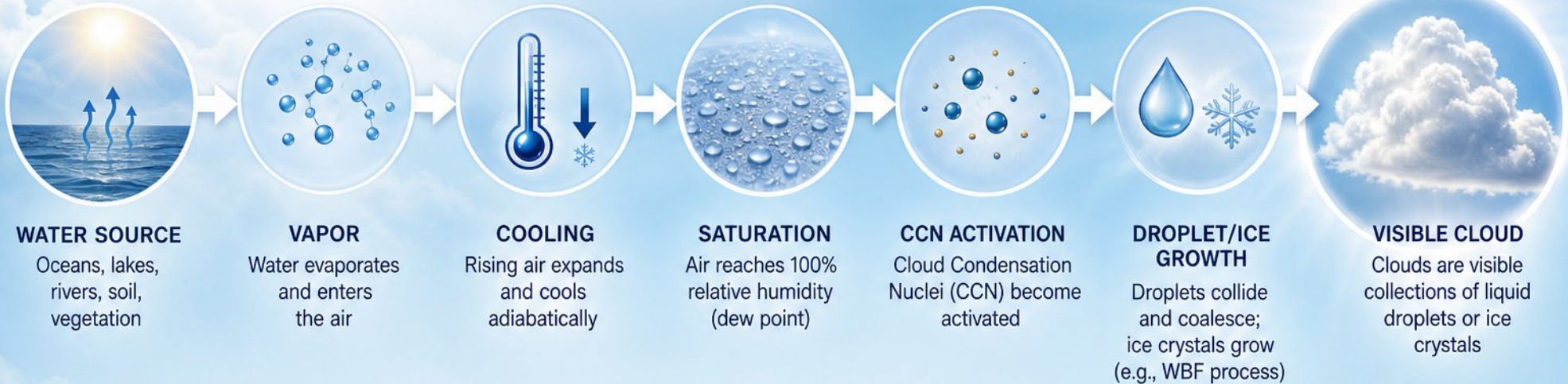
INVISIBLE

INVISIBLE TO VISIBLE SPECTRUM



VISIBLE

# The Full Chain: Linking Vapor, Cooling, Nuclei, and Cloud Types



- Water source → vapor → cooling → saturation → CCN activation → droplet/ice growth → visible cloud



- Key terms: relative humidity, dew point, adiabatic cooling, CCN, LCL, supercooled water, WBF process



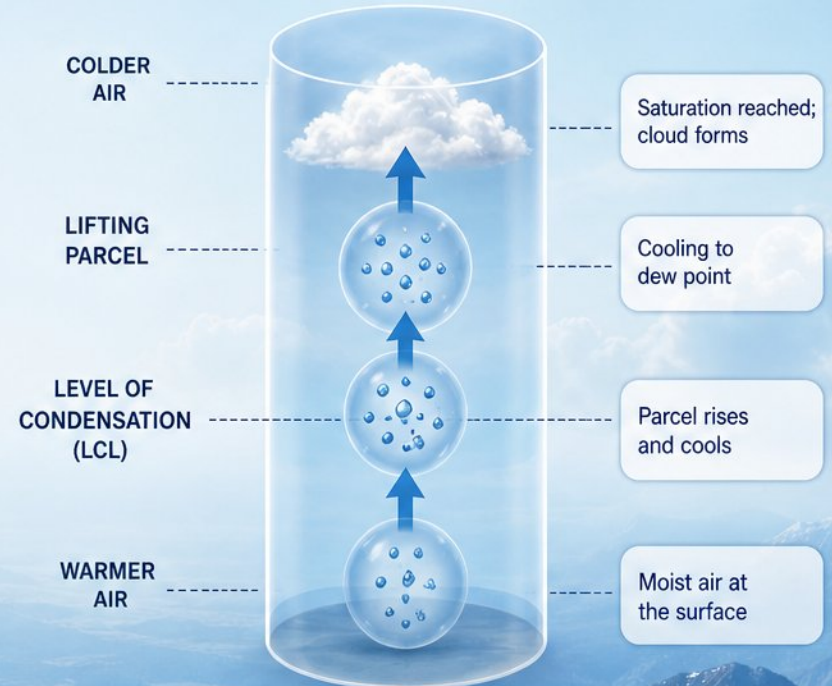
- Clouds are liquid droplets or ice crystals, not water vapor



- Condensation releases latent heat, warming the parcel



- Different cloud types form under different stability, moisture, and updraft conditions



# Summary and Beyond: From Clouds to Weather



- Clouds require water vapor, cooling, and condensation nuclei



- Use these three pillars to explain any cloud type



- Next: droplet growth, precipitation, and climate impacts



- Every cloud reveals invisible atmospheric processes



INVISIBLE

INVISIBLE TO VISIBLE

VISIBLE



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